# **Study of Nuclear Properties of Even-Even** (164,166 and 168) W **Isotopesin the Interaction Boson Model (IBM-1)**

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**Abstract:** This study presents a comprehensive analysis of tungsten isotopes (W) with atomic numbers ranging from 164 to 168 using the Interaction Boson Model (IBM-1). The properties and structures of these isotopes were investigated by considering three dynamic symmetries: rotational SU (3), O (6), and dynamic vibrational SU (5). By comparing practical energy ratios with ideal values for each dynamic symmetry, the classification of the isotopes in transitional regions was determined. Specifically, isotopes 164-166 were found to belong to the SU (3)-O (6) transitional region, while isotope 168 adhered to the SU (3)-O (6) region. The study computed the Hamiltonian function and quadrupole electric transition effect and compared theoretical and experimental energy levels. The determination of the number of valence bosons aided in characterizing the atomic composition of the studied isotopes. These findings enhance our understanding of the nuclear properties of tungsten isotopes within the framework of IBM-1, providing insightsinto their dynamic symmetries and energy levels.

**Keywords:** Interacting boson model-1, even-even isotopes, Energy level, w isotopes

# *Introduction*

# *1.1. Applications and uses of tungsten*

Tungsten is a chemical element with unique properties and properties, and it is widely used in nuclear applications. Here are some of the main advantages and properties of tungsten and its uses in nuclear applications:

1. High Heat Resistance: Tungsten has a high melting point of about 3422°C, which makes it ideal for use in hightemperature applications such as nuclear cores.

2. High Hardness: Tungsten is one of the hardest metals known, and this makes it resistant to abrasion, wear, and chemical corrosion, making it suitable for use in tough environments and high tolerance requirements.

3. High Density: Tungsten has a density as high as 19.3  $g/cm<sup>3</sup>$ , which helps in increasing the amount of radiation scattering materials in nuclear applications.

4. Radiation resistance: Tungsten has excellent radiation tolerance and is therefore used in manufacturing cooling shafts and covers for nuclear power reactors.

5. High corrosion resistance: Thanks to its high hardness and corrosion resistance, tungsten is used in the manufacture of pipe parts and other devices used in nuclear power plants[1].

for electricity. There are many other uses for tungsten in nuclear applications, including:

1. Radiator control rods: Tungsten is used in the manufacture of Control Rods that are used in nuclear power reactors to control the speed of nuclear reaction and adjust the proportion of radiated neutrons.

2. Fuel-carrying filaments: Tungsten is used as a nuclear fuel carrier in some types of nuclear reactors. Tungsten-carrying filaments provide the advantage of a high fuel density and help improve the efficiency of nuclear reactors.

3. Fuel Pens: Tungsten is used in the manufacture of nuclear fuel pens that contain nuclear fuel, such as radioactive uranium (Uranium-235). Tungsten fuel pens enhance the efficiency of nuclear energy recovery and act as a barrier to protect against nuclear material leakage [2].

4. Nuclear bomb electrodes: Tungsten is used in the manufacture of nuclear bomb electrodes, as it directs and rations the flow of nuclear energy and reduces heat loss anddispersion.

5. Radiation detection devices: Tungsten is used in the manufacture of nuclear radiation detection devices, such as nuclear spectrometers and mass radiation detectors. Tungsten's advantages such as its high radiation resistance and hardness help to increase the accuracy and reliability of these devices.These are some of the main uses of tungsten [3].



# *1.2. Isotopes and their types in tungsten*

Tungsten has six stable isotopes and several unstable isotopes. Here is a list of the stable isotopes of tungsten and the relative concentrations of each isotope[4]:

- 1. Tungsten-180 (180W) The most abundant isotope and makes up about 0.12% of natural tungsten.
- 2. Tungsten-182 (182W) makes up about 26.5% of naturaltungsten.
- 3. Tungsten-183 (183W) makes up about 14.31% of naturaltungsten.
- 4. Tungsten-184 (184W) makes up about 30.64% of naturaltungsten.
- 5. Tungsten-186 (186W) makes up about 28.43% of naturaltungsten.
- 6. Tungsten-188 (188W) makes up about 1.62% of naturaltungsten.

The investigation of energy levels in isotopes plays a crucial role in unravelling the complexities of nuclear structure. In this work, we employ the IBM-1 model to analyze the energy spectra of isotopes, thereby shedding light on their underlying symmetries. By comparing the calculated energy levels with practical values, we ascertain the efficacy of the IBM-1 model in describing the behavior of these isotopes. Additionally, we explore the impact of the number of valence bosons on the energy levels and emphasize the significance of understanding the proportional height variation among these levels.

In addition to the stable isotopes, there are several unstable isotopes of tungsten that undergo radioactive decay into other elements. Some unstable isotopes of tungsten include tungsten-179, tungsten-181, tungsten-185, tungsten-187, tungsten-189, tungsten-190, tungsten-192, and others.Unstable isotopes have different half-lives and differ in patterns of radioactive decay which can include alpha decay, beta decay, gamma decay, and neutron decay. Researchers in the field of nuclear physics continue to be intrigued by the atomic structure problem, seeking a comprehensive model that can provide a holistic understanding of nuclear behavior. This monograph explores the IBM-1, proposed by Arima and Iachello in 1974, as a promising approach towards achieving this goal. By addressing the need for a unified model, the IBM-1 offers valuable insights into the intricate nature of nuclear systems. This work aims to present a concise overview of the IBM-1, highlighting its potential applications and contributions to nuclear physics.[5], This study presents an appropriate framework to elucidate the behaviour of heavy and medium atomic nuclei. The proposed model considers the importance of low collective states in even nuclei, which possess a finite number of protons or neutrons close to the magic numbers (2, 8, 20, 28, 50, 82, 126). These collective states play an important role in the excitation of valence protons (neutrons). Moreover, the distribution properties of the lower states are determined by treating pairs of similar molecules as a single construct. Specifically, bosons with zero angular momentum  $(i = 0)$  are referred to as S-type bosons (Sπ (SV)), while those with angular momentum (I = 2) are known as D-type bosons (Dπ (DV)). This approach sheds light on understanding nuclear structure and phenomenology, emphasizing the importance of collective states and boson therapy. (I=0) are called s-type bosons and defined as  $s<sub>\pi</sub>(s<sub>v</sub>)$ , whereas those have angular momentum (I=2) are called dtype bosons and defined as  $d_{\pi}(d_{\nu})$ [6–9]. Taking into consideration the number of the valence pairs of protons (neutrons) defined by  $N_n( Nv)$ , which are calculated to the nearest closed shell. If the shell is less than half full, the boson number is particle pairs, and if it is more than half full, the boson number is holes type. For example, the isotope (W-174) has (74) protons, and the nearest closed shell to it is (82), so the number of protons bosons N<sub>π</sub>=  $\frac{82-74}{2}$ , which is type of pairs of

holes, and the number of bosons of neutrons  $Nv = \frac{1}{2}$ . W-174 has 100 neutrons, which are of particle pairs type. The first interactive boson model (IBM-1), which was used in this research, does not distinguish between proton bosons

and neutron bosons, so its treatment depends on the total number of bosons, as the N=N<sub>n</sub>+N<sub>v</sub>. The decay of excited nuclear levels leads to the emission of electromagnetic gamma rays[10,11], and according to the selection rules for nuclear spin and symmetry, the type of radiation emitted is determined Whether it is from an electric or magnetic multipolar or both.

#### *Theoretical foundations of IBM-1*

In the IBM-1 model, the Hamiltonian operator can be expressed as follows[12-15]:<br> $\hat{a}$  is sure as  $\hat{a}$  is  $\hat{a}$  is

*H* = *s ns* + *d nd* + *a p* . *p*+ *a L*. *L*+ *a Q* .*Q* + *a T* .*T* + *a T* .*T* + (1) 0 1 2 3 3 3 4 4 4

Where  $\varepsilon_s$  and  $\varepsilon_d$  represent (s-boson) & (d-boson), energies, respectively.

While the operators  $\stackrel{\wedge}{P},\stackrel{\wedge}{Q},\stackrel{\wedge}{L},\stackrel{\wedge}{T}{}_3$  ,  $\stackrel{\wedge}{T}{}_{4}$  are given by the following relations:

Where:

 $\varepsilon$  is the boson energy,  $\varepsilon = \varepsilon_d - \varepsilon_s$  (1-a)

 $n_d$  is the total number of d boson operator,  $n_d = (d^\dagger)$  $(1-b)$ 

*p*, is the pairing operator  $p = \frac{1}{2}[(d \cdot d) - (s \cdot s)]$  (1-c)

L, is the angular momentum operator  $L = \sqrt{10} [(d^{\dagger} \times d)]$  (1-d)

Q, is the quadrupole operator  $Q = [d^{\dagger} \times s + s^{\dagger} \times d]^2 + \chi[d + \times d]^2$ (1-e)

(x) is the quadrupole structure parameter and it takes the values 0 and  $\pm \sqrt{7/2}$ ,

 $T_r$  is the octupole operator for (r = 3) and hexadecapolar operator for (r = 4).

 $\alpha_0$ ,  $\alpha_1$ ,  $\alpha_2$ ,  $\alpha_3$  and  $\alpha_4$  are the parameters Indicate the force of the pairing, angular momentum, quadrupole,

octupole and hexadecapole interaction between the bosons.

 $\hat{\rho}$  pairing operator,  $\hat{\pmb{\varrho}}$  quadruple momentum operator,  $\hat{\imath}$  :The angular momentum,  $\hat{\pmb{\eta}}$ : Octupole operator, hexadecapole operator.  $\hat{T}_4$ :

There are only three cases for the U (6) group possible to solve analytically in the IBM-I model, These are anointed the series of vibrational group U (5), the series of transitional group O(6), and the series of the rotational group SU(3). In

nuclei that are within symmetry SU(3), the coefficients  $\frac{a}{1}$  and  $\frac{a}{2}$ , and associated with the two parameters  $\frac{a}{1}L^2$  and

 $a_2Q^2$  , , respectively, are more important and have an impactful effect, i.e., these two coefficients govern the energy spectrum in these type of nuclei.

Values of  $a_1$  and  $a_2$  can be calculated from the following relations[16,17].<br>
2*a*  $E^+$ 

$$
a_{1} = \frac{2a_{2} + \frac{E_{+s}}{2^{s}}}{8 - 6}
$$
\n
$$
a_{2} = \frac{E_{2^{*}g} - E_{-s}}{3(2N - 1)}
$$
\n(3)

where:  $E_{\lambda^+}$  energy level in  $\beta$  band,  $E_{\lambda^+}$ energy level in  $\beta$  band,  $E_{2^+g}$  energy level in  $g$  band, and N is the total number of valance bosons.

When choosing the values of the coefficients in the Hamilton equation, we must know some prior background information about the nucleus under investigation. For instance, the value of R4 ratio for the ground state band[18,19], which equal  $(E_{\frac{4}{4}}/E_{\frac{4}{2}})$ , the value of R<sub>4</sub> ratio differ from symmetry to another. where its value in the vibration

symmetry U (5), lies in the range  $2 \le RA \le 2.4$ , while in the transition symmetry O (5) it is limited in the span 2.4  $\le RA \le$ 3, and in the rotational symmetry it is constrained in between  $3 ≤ R4 ≤ 3.33$ .

in light of the above discussion, Certain parameters can be weighed over others in the operator of the Hamiltonian function equation[1].

 $\wedge$ 

# **RESULTS AND DISCUSSION**

The special thing about this study is that it analyzes the properties and structures of tungsten isotopes using the interaction boson model (IBM-1) and considers three types of dynamic symmetries: rotational dynamic symmetry SU (3), translational dynamic symmetry O (6), and dynamic vibrational symmetry SU (5). It compares the practical energy ratios of the tungsten isotopes with the ideal values for each dynamic symmetry and determines that the isotopes (164- 166) belong to the SU (3)-O (6) transitional region, while the 168 isotope adheres to the SU(3)-O (6) region. The study also computes the Hamiltonian function and quadrupole electric transition effect, compares theoretical and experimental energy levels, and determines the number of valence bosons in the studied isotopes.

The W isotopes with atomic number ( $Z = 74$ ) and mass number ( $A = 164-168$ ) were studied using the interaction boson model, IBM-1. Most of the interaction models studding the properties and structures of nuclei, divides them into three types of dynamic symmetries: rotational dynamic symmetry SU (3), translational dynamic symmetry (O(6), and dynamic + +

vibrational symmetry (SU(5)). The energy ratios play a role in determining dynamic symmetries. E ( + 4  $/_{2+}$ ), E(  $8^{^{+}}/_{2}$ +), E(  $8^{^{+}}/_{2}$ +),

 $E(\frac{7}{2}+1)$ Based on our theoretical calculations, these are the ideal amounts of energy for all three dynamics. Asymmetric Dynamic Symmetry in Rotation SU (3(

 $E(4^+)$   $E(6^+)$   $E(8^+)$  $\left[\frac{3}{2} \cdot \frac{3}{2} \cdot 3\right]$ ,  $\left[\frac{3}{2} \cdot \frac{3}{2} \cdot 3\right]$  ,  $\left[\frac{3}{2} \cdot \frac{3}{2} \cdot 3\right]$ 

Dynamic Transition Symmetry O(6)<br>  $\frac{\mathbb{E}^{(4^+)}}{\mathbb{E}^{(4^+)}} = 2.5 \quad \int_{E(2^+)}^{E(6^+)} = 4.5 \quad \int_{E(2^+)}^{E(8^+)} = 7 \quad .$ Dynamic Symmetry of Vibrations SU (5)

$$
\begin{aligned}\n\begin{bmatrix}\nE(4^+) \\
\overline{E(2^+)}\n\end{bmatrix} & E(6^+) \\
\overline{E(2^+)} & = 3 \quad \overline{E(2^+)} = 4\n\end{aligned}
$$

The dynamic symmetries overlap, resulting in three transitional regions: SU (5)-SU (3), Where there is a convergence of the rotational and vibrational dynamic symmetrical structures. Following this transitional region, nuclei exhibit both rotational and vibrational motion behavior. SU (3)-O (6) the nuclei here exhibit a combination rhythm of rotational and transitional motion. SU(5)-O(6) Where vibration and change meet in this transitional zone[20]. This transitional region's nuclei exhibit behavior that is a mix of vibrational and transitional motions. The practical energy ratios were used to

+ + + 8 + determine the dynamic symmetry of the even-pair tungsten isotopes reported in this paper[21-23]. E ( $\gamma_{2+}$ ) E(<br>  $\epsilon^+$  $6^+$  4<sup>+</sup> E(4<sup>+</sup>)  $\gamma_{2+}$ ) E(  $\gamma_{2+}$ ) .The ideal values were as follows:  $\mathbb{I}_{E(2^+)}$  The mass setting on the W isotope synthesis process is equal to (164,166,168) ,(2.5,2.6,2.8) As well as value, respectively. On the other hand,  $\mathbb$ practical value of 4.3,4.87,5.2 respectively.  $\mathbb{E}^{(8^+)}_{E(2^+)}$  On the other hand, have practical values 6.4,7.4,8.03 Respectively. The nuclei can be inferred from these values. (164-166) To what's pure. SO (6) Between the two regions.  $\mathbb{E}^{\frac{E(4^+)}{E(2^+)}}$  and

 $+$ <sup>1</sup>  $+$ <sub>E</sub> $(2$  +  $\mathbb{I}_{E(2^+)}^{E(6^+)}$  and  $\mathbb{I}_{E(2^+)}^{E(8^+)}$  This transition symmetry is identical or very close to the ideal values, and the nuclei (168) adhere

+ + + to the transition region. SU (3)-O (6) The greater the value<br>  $\frac{E(4^+)}{E(2^+)}$  and  $\quad \frac{E(6^+)}{E(2^+)}$  and  $\quad \frac{E(8^+)}{E(2^+)}$  Rotational dynamic symmetry SU (3) is preferable, but it can be achieved by increasing the mass number A to achieve translational active symmetry.

O (6) instead. The Tungsten's (A=166-164) isotopes have different energy levels Were computed utilizing the Hamilton function effect of transition symmetry O (6) equation[12,15],:

$$
\hat{H}_{III} = a_o \hat{P} + \hat{P} + a_1 \hat{L} \cdot \hat{L} + a_3 \hat{T}^3 \cdot \hat{T}^3
$$
 (4)

The circular region equation is solved using the Hamilton function effect[12,15]:

$$
\stackrel{\wedge}{H} = a_0 \stackrel{\wedge^+}{P} \stackrel{\wedge}{P} + a_2 \stackrel{\wedge}{Q} \stackrel{\wedge}{Q} \tag{5}
$$

to determine the168 isotopes' energy levels. The Hamilton effect symmetry is applied to the rotations equation [12,15]:  $\wedge$   $\wedge$ 

$$
H (II) = a_1 L \cdot L + a_2 Q \cdot Q \tag{6}
$$

to compute the isotopes' energies. Information about the isotopes' energy levels can be gleaned from tables like (1-3), which detail the Hamilton function's influence on those levels.

### *Proportions of energy levels:*

Tungsten's even-even (A=164-168) isotopes were theoretically calculated by IBS1 package. The theoreticalresults derived from scientific literature and periodicals were compared. These isotopes' energy bundles are organized into tables (2) based on theoretical and experimentally derived values. The statisticalscale RMSD was used to compare the actual values with the predicted values for each energy bundle [24].

$$
RMSD = \left[ \frac{1}{N} \sum (E_{cal} - E_{Exp})^2 \right]^{\frac{1}{2}}
$$
 (7)

The RMSD values in these tables are interesting to note W168 (0.015) for the ground state beam; W166 (0.02); and W166 (0.02), as well as (0.011) for the 166 W ground state beam. Small and close-to-zero RMSD values were observed in the data set., The isotopes 168, W166, and W164 have no practical value in the bands. The number of valence bosons in the isotopes being studied is the first step in determining the Hamilton function's coefficients. The isotopes' nuclei W164, 166, and 168 concerning their atomic composition Z =74 As a result, the number of proton-boson valence bosons equals the number of filled shells closest to the proton Nπ=4,4,4. Due to the lack of filling in the outer shell, there are holes in each section. As they are closer to the filled surface, neutrons can be found in their nuclei Thus, the number of bosonneutrons is equal to Nυ=4,5,6 Accordingly, the total number of valence bosons can be calculated N =.8,9,10 A row of bosons, one after another. The first criterion by which the parameters of the Hamilton function used in our research are determined and controlled is the matching principle between the theoretically calculated energy levels in the IBM1 model and their practically measured values Table 1 shows that: Coefficients for the isotope-specific Hamiltonian functions for isotopes W 164,166 and 168 It is shown in the table belw:

**Table 1:that using the (IBS1) and (IBMT), the Hamiltonian function and the quadrupole electric transition effect of the even-even tungsten isotopes W(A=164–168)**

			$W(A-10+100)$							
Isotopes	Behavior	EPS )Mev(	a <sub>0</sub> )Mev(	a1 )Mev)	a2 )Mev(	a3 )Mev)	a4 )Mev)	$\chi$	$\alpha$ 2 (e <sub>b</sub> )	$\beta$ (eb)
$\frac{164}{74}$	O(6)	0.000000	0.058700	0.013000	$-0.000000$	0.182200	0.000000	$-0.000000$		п.
$M_{\rm QD}^{bL}$	O(6)	0.000000	0.028700	0.016100	-0.000000	0.110500	0.00000	$-0.000000$		
$M_{\rm QD}^{bL}$	(9)0-(8)ns	0.000000	0.000420	0.015900	$-0.023700$	0.013600	0.000000	$-0.650000$	$\blacksquare$	

Table (1) shows the Hamiltonian function parameters and the quadrupole electric transition effect of the even-even tungsten isotopes W(A=164 – 168) using IBS1.For and IBMT For The data in Table 1 was used to figure out the theoretical energy levels, which were found to be In the IBM-1 model, the W 164 tungsten isotope is a good match for all energy beams in terms of practical values. The value of R4 was found to be equal to when the theoretical energy levels were calculated, and A kernel forthe IBM 1-model is expected. For example, isotope 166-164 isotope tracesthe pure transition region, SO (6) the pattern is oscillatory or transitional. The rotational pattern of 168 follows the SU (3)–O (6) region. Despite some minor diffraction or a small increase in energy values at the beam, the energy spectrum of the W 166 nucleus was found to be a good match with the practical spectrum γ. Since the nucleus of this isotope belongs to definition SO (6), we can say that it has the properties of a vibratory group In addition, the low-energy spectrum of the W 168 isotope is very similar to the practice range.It was found that the isotopes that were dealt with carried close nuclear shapes through this very circle. It was found that the w isotopes can be described as distorted nuclei, that is, they do not belong to any of the three pure dynamic symmetry[25], but they are those that tend strongly to symmetry SU(3) and possess to a small degree the characteristics of the group or Determination O(6) and we see this clearly when

entering the parameters into the energy spectrum calculations[11,26] and this is what causes the symmetry breaking SU(3) [27]in the direction of O(6).





Table (2) shows that the tungsten isotope with mass number (W-164) is a stable isotope and the energy levels in the ground state of this isotope are formed by neutron transitions. The distribution of neutrons in the electronic layers varies according to different nuclear models and possible neutron interactions. The isotope tungsten-164 has no beta or gamma energy due to its stable nuclear structure. The isotope tungsten-164 has 76 protons and 88 neutrons, which gives it a stable nuclear configuration. Table (2) shows the presence of several protons and neutrons in the nucleus affects the stability of the isotope, and isotopes that contain different numbers of protons and neutrons can be unstable and release beta or gamma energy during nuclear decay. However, the isotope of tungsten-164 may be able to decay by alpha decay, double beta decay, or some other decay process, but these processes are not very common to tungsten-164.It should be noted that the nuclear properties and interactions of isotopes depend on their specific nuclear composition and on other factors related to energy and nuclear enhancers.From the observation of the results listed in Table 2, it was found that the values of the theoretical energy levels calculated in the IBM-1 model for w isotope match well with the practical values for all energy bands (g, β, γ).When calculating the value of RMSD for the calculated theoretical energy levels, it was found that it is equal to 0.096, so it lies within the same range that goes back to the determination SU (3), which is very close to the practical value. The resulting energy spectrum with the practical spectrum despite some simple diffraction or a small rise in energy values at the γ-beam, that is, which shows the analog determination and to know its type and belong to this nucleus. It is located within the same range, and thus we can say that the nucleus of the isotope w belongs to the definition SU (3), that is, it bears the properties of the group.





**Fig 1(a)** shows the relationship between the energy on the vertical axis and the magnetic angular moment on the horizontal axis.The finding illustrated in Fig 1a , Fig 1b the relationship between the beam energy was drawn in the form of horizontal lines in terms of the geometric variables β and γ for the W164 isotope. Fig.2 b, The relationship of the total angular momentum to the energy levels of the theoretical beams ( g) computed using (IBM-1 ) and their comparison with the practical values of the isotope W164 When drawing the relationship between the angular momentum and energy, it was found that there is a match between the practical values given by the International Atomic Energy Agency and the values calculated by the model of interacting bosons and an excellent match was found in most of the linear points.





Table 3 shows the energy levels obtained in the current research in relation to the mass number and compared with the practical values of these isotopes, which were in good agreement with them. The variation in the proportional height of these levels is because the isotopes belong to one nucleus. therefore, the numbers of valance bosons for these isotopes are N<sub> $\pi$ </sub> boson, = 4, N<sub>v</sub> = 5, N =9 respectively. When studying this isotope under IBM-1, we obtain the energy spectra for these isotopes as shown in table (3) and we have a good agreement with experimental energy. From the ratio RMSD for calculated (Theoretical) energy state, we have obtained a good agreement value with the experimental results from a point of view the symmetry limits.

In addition to the remarkable agreement between the calculated energy spectra and experimental values, further insights can be gained by analyzing the symmetry limits in the studied isotopes. The investigation of these isotopes under the framework of the Interacting Boson Model-1 (IBM-1) has yielded significant results. Table 3 presents the energy levels obtained in our current research, which are directly related to the mass number of each isotope. It is noteworthy that the observed variation in the proportional height of these energy levels is primarily attributed to the isotopes belonging to a single nucleus. The valance boson numbers for the respective isotopes are determined as  $N\pi$  boson = 4, Nv boson = 5, and N = 9. These values have played a crucial role in shaping the energy spectra for these isotopes, as depicted in Table 3. The calculated (theoretical) energy states exhibit excellent agreement with the experimental results, as evidenced by the ratio of Root Mean Square Deviation (RMSD) values. This further emphasizes the validity of the theoretical approach utilized in our study. By exploring the symmetry limits inherent in these isotopes, our research contributes to a deeper understanding of their nuclear structure and behaviors. The obtained agreement between theory and experiment not only validates the IBM-1 framework but also highlights its effectiveness in capturing the essential features of the studied isotopes. These findings open up new avenues for investigating nuclear systems and pave the way for future explorations in the field.



Fig.2:: a) A comparison of theoretical (g.  $\gamma$ .  $\beta$ ) beam energies (P.W) to their actual, b) The relationship of the

In addition to the findings depicted in Fig. 2(a, b), the study also investigated the impact of varying geometric variables  $β$ and γ on the relationship between beam energy and the magnetic angular moment for the W166 isotope. Notably, when plotting this relationship, horizontal lines were observed, indicating a consistent energy distribution across different values of β and γ. Furthermore, Fig. 2(b) explored the relationship between the total angular momentum and energy levels of the theoretical beams, as computed using the IBM-1 (Interacting Boson Model) framework. These theoretical values were then compared to the practical values obtained from the International Atomic Energy Agency (IAEA) for the W166 isotope. Impressively, a remarkable alignment was observed between the calculated IBM-1 values and the practical measurements across a significant number of data points, signifying a high level of agreement. This agreement between the theoretical and practical values substantiates the efficacy of the IBM-1 model in accurately predicting the angular momentum and energy levels of the W166 isotope. The study's findings support the use of this model for future investigations in nuclear physics, demonstrating its potential in advancing our understanding of the underlying principles governing atomic nuclei. These results present a significant contribution to the field and warrant further exploration and validation in the pursuit of more comprehensive and accurate atomic models.





Table 4 shows the energy levels obtained in the current research in relation to the mass number and compared with the practical values of these isotopes, which were in good agreement with them. The variation in the proportional height of these levels is since the isotopes belong to one nucleus. therefore, the numbers of valance bosons for these isotopes are  $N_{\pi}$  boson, = 4,  $N_{\nu}$  = 6, N = 10 respectively. When studying this isotopes under IBM-1, we obtain the energy spectra for these isotopes as shown in table (4) and we have a good agreement with experimental energy spectra as shown in figures (Fig.3) . From the ratio RMSD for calculated (Theoretical) energy state, we have obtained a good agreement values with the experimental results from a point of view the symmetry limits. Our analysis focuses on the variation in the proportional height of these energy levels, which arises from the isotopes belonging to a single nucleus. Furthermore, we determine the numbers of valence bosons for each isotope as  $N\pi$  boson = 4, Nv boson = 6, and N = 10. The calculated energy spectra using the IBM-1 model are presented in Table 4 and are found to be in excellent concurrence with the experimental energy spectra, as demonstrated in Figure 3. The Root Mean Square Deviation (RMSD) ratios of the calculated theoretical energy states indicate a strong agreement with the experimental results, particularly with regard to the symmetry limits. This study provides valuable insights into the energy behavior and symmetrical properties of isotopes, contributing to our understanding of nuclear structure. By employing the IBM-1 model, we have successfully captured the energy behaviors of isotopes and demonstrated its agreement with experimental data. The observed symmetry limits, as evidenced by the RMSD ratios, further substantiate the effectiveness of the IBM-1 model in describing the nuclear structure. This study contributes to our understanding of the interplay between valence bosons, energy spectra, and symmetries in isotopes, emphasizing the utility of the IBM-1 model in nuclear physics research.



Fig.3: a) A comparison of theoretical (g. γ . β) beam energies (P.W) to their actual, b) The relationship of the total angular

in Fig. 3(a,b), the study also investigated the impact of varying geometric variables β and γ on the relationship between beam energy and the magnetic angular moment for the W168 isotope. Notably, when plotting this relationship, horizontal lines were observed, indicating a consistent energy distribution across different values of β and γ. Furthermore, Fig. 3(b) explored the relationship between the total angular momentum and energy levels of the theoretical beams, as computed using the IBM-1 (Interacting Boson Model) framework. These theoretical values were then compared to the practical values obtained from the International Atomic Energy Agency (IAEA) for the W168 isotope. Impressively, a remarkable alignment was observed between the calculated IBM-1 values and the practical measurements across a significant number of data points, signifying a high level of agreement. This agreement between the theoretical and practical values substantiates the efficacy of the IBM-1 model in accurately predicting the angular momentum and energy levels of the W168 isotope. The study's findings support the use of this model for future investigations in nuclear physics, demonstrating its potential in advancing our understanding of the underlying principles governing atomic nuclei. These results present a significant contribution to the field and warrant further exploration and validation in the pursuit of more comprehensive and accurate atomicmodels.

# *CONCLUSIONS*

In the study focused on the isotopes of tungsten with atomic number 74 and mass numbers 164-168. The interaction boson model, IBM-1, was used to analyze the properties and structures of these nuclei. Three types of dynamic symmetries were considered: rotational dynamic symmetry SU(3), translational dynamic symmetry O(6), and dynamic vibrational symmetry SU(5).The energy ratios E(8+⁄2+), E(6+⁄2+), and E(4+⁄2+) were used to determine the dynamic symmetries. The ideal energy ratios for each dynamic symmetry were provided:Asymmetric Dynamic Symmetry in Rotation SU(3):E(4+)/E(2+) = 3.33,E(6+)/E(2+) =,E(8+)/E(2+) = 12.Dynamic Transition Symmetry O(6):E(4+)/E(2+) = 2.5,E(6+)/E(2+) = 4.5,E(8+)/E(2+) = 7.Dynamic Symmetry of Vibrations SU(5):E(4+)/E(2+) = 2,E(6+)/E(2+) = 3,E(8+)/E(2+) = 4.The nuclei in the transitional regions exhibit a mix of rotational, vibrational, and transitional motion behaviors. The practical energy ratios for the tungsten isotopes (164, 166, and 168) were compared to the ideal values, and it was found that (164-166) isotopes belong to the SU (3)-O (6) transitional region, while the 168 isotope adheres to the SU(3)-O(6) region. The Hamiltonian function and the quadrupole electric transition effect of the even-pair tungsten isotopes were computed using the IBM-1 model. The energy levels of the isotopes were determined, and a comparison was made between theoretical and experimental values using the root mean square deviation (RMSD). The RMSD values were close to zero for the ground state beams of W166 and W164 isotopes, indicating a good agreement between the predicted and measured energy levels. The number of valence bosons in the studied isotopes was determined based on their atomic composition. For W164, W166, and W168, the number of proton-boson valence bosons (Nπ) was equal to 4, and the number of boson-neutrons (Nυ) was equal to 4, 5, and 6, respectively. The total number of valence bosons (N) was calculated as 8, 9, and 10 for W164, W166, and W168, respectively. The coefficients for the isotope-specific Hamiltonian functions were determined based on the IBM-1 model and the matching principle between theoretically calculated energy levels and experimental measurements. In conclusion, our study provides compelling evidence of the successful application of the IBM-1 model in predicting and explaining the energy spectra of these isotopes. The agreement with experimental data and the insights gained from studying the symmetry limits underscore the significance of our findings. This work contributes to the existing body of knowledge in nuclear physics and strengthens the foundation for further research and advancements in thefield.

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